

# TECHNICAL INFORMATION

VOCs Free Low Solids No-clean Flux

JS-3000V-3

## 1. Features

- Extremely low residue.
- Designed for soldering with Sn-Ag, Sn-Ag-Cu, Sn-Cu.
- Suitable for spray application.

## 2. Specifications

Item		<b>JS-3000V-3</b>	Remark
Specific gravity		1.015	--
Solids content %		5.0	--
Acid value KOHmg/g		35.6	Titration
Halogen content		0	Potentiometric titration
Flux type		ORL0	ANSI/J-STD-004
Silver chromate paper test		No discoloration	MIL-F-14256E
Copper plate corrosion test		No evidence corrosion	40°C × 95%RH × 96Hr
Copper mirror corrosion test		Partial breakthrough	IPC-TM-650 2.3.32 (at IPA solution)
Water extract resistivity Ω • m		$> 1 \times 10^2$	MIL
Solder spreadability %	Sn-37Pb	$> 85$	150°C × 60sec. Oxide copper plate
	Sn-3.0Ag-0.5Cu	$> 75$	
Surface insulation resistance (Ω)	Initial value	$> 1 \times 10^{12}$	After soldering (250°C×4s)
	In85°C/85%RH/168Hr	$> 1 \times 10^9$	Measured in thermohygrostat
	After 85°C/85%RH/168Hr	$> 1 \times 10^{12}$	Out of thermohygrostat
Voltage applied Insulation resistance (Ω)	Initial value	$> 1 \times 10^{12}$	After soldering (250°C×4s)
	In 85°C/85%RH/168Hr	$> 1 \times 10^9$	Measured in thermohygrostat
	After 85°C/85%RH/168Hr	$> 1 \times 10^{12}$	Out of thermohygrostat
	Electromigration	No evidence of electromigration	--
Wettability (Meniscograph)	Wetting speed Sec.	0.57	Polished copper plate Solder: Sn-3.0Ag-0.5Cu
	H <sub>4</sub> tensile strength mN/m	388	
	Wetting speed Sec.	0.69	Polished nickel plate Solder: Sn-3.0Ag-0.5Cu
	H <sub>4</sub> tensile strength mN/m	315	
Dryness of flux residue		No attachment of chalk powder	250°C × 5sec

### 3. Specific gravity

The test shall be carried out to determine the specific gravity of liquid flux by using the float hydrometer standardized in JIS-B-7525.

Take the sample flux into a cleaned and dried glass tube and put it in a constant temperature bath of temperature 20°C.

Floating the specific hydrometer in the sample flux in the glass tube, measure the specific gravity by reading upper edge of meniscus line.

[ Result ]

Average	1.015
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### 4. Solids content

The flux shall be sampled approx. 10g and weighted (W1). After heating at 105±2°C for 5 hours, measure the weight again (W2).

$$\text{Solids content (wt\%)} = \frac{\text{Weight after heating (W2; g)}}{\text{Weight before heating (W1;g)}}$$

Repeat the test twice and take an average.

[ Result ]

n	1	5.0
	2	5.0
Average (%)		5.0

### 5. Acid value

This test shall be carried out to determine the acid value in the liquid flux by the manual titration method.

Put 2 gs of flux in the precision of 1/1000g into the beaker of 200ml and pour approx. 50ml of ethylalcohol/benzene solution (1:2) or isopropyl alcohol, and drop 2~3 drops of phenolphthalein indicator to obtain the sample.

Titrate it with N/2 potassium hydroxide/ethyl alcohol standard solution until the end point is obtained, where the sample color turns to pale red from colorless and remains red for more than 30 seconds. Conduct the blank test through the entire process and calculate the acid value.

$$\text{Acid value [KOHmg/g]} = \frac{28.5 \times (A-B) \times F}{S}$$

- A : Quantity (ml) of N/2 potassium hydroxide/ethylalcohol standard solution used this test.  
 B : Quantity (ml) of N/2 potassium hydroxide/ethylalcohol standard solution used for blank test.  
 F : Factor of N/2 potassium hydroxide/ethylalcohol standard solution.  
 S : Quantity (g) of sample.

[ Result ]

n	1	35.6
	2	35.6
Average (%)		35.6

## 6. Halogen content (Chloride content)

This test shall be carried out to determine the halogen content in liquid flux by the electric potentiometric titration method.

Put approx. 5gs of flux into the beaker of 200ml and weigh it with the balance in the precision of 1/100gs and pour approx. 100ml of isopropyl alcohol to obtain the sample.

Transfer the sample to the electric potentiometric titration equipment and titrate it with 1/50N silver nitrate standard solution by stirring it with a magnetic stirrer until the end point where electric potential changes largely is determined.

Carry out the blank test through the entire process and calculate the halogen content in the flux from following formula.

Repeat the test twice and take an average.

$$\text{Halogen content (\%)} = \frac{(A-B) \times 0.000709 \times f}{\text{Mass of flux (g)}} \times 100$$

- A : Amount (ml) of 1/50N silver nitrate solution used for the entire test  
 B : Amount (ml) of 1/50N silver nitrate solution used for the blank test  
 0.000709 : Amount (g) of halogen corresponds to 1ml of 1/50N silver nitrate solution  
 f : Factor of 1/50N silver nitrate solution

[ Result ]

n	1	0
	2	0
Average (%)		0

## 7. Silver chromate paper test

Place one drop of test flux on each piece of silver chromate test paper specified in MIL-F-14256E. Allow the droplet to remain on each test piece for a minimum of 15 seconds. After 15 seconds, immediately immerse each test paper in clean isopropyl alcohol to remove residual organic materials.

Allow each test paper to dry for minutes, then examine for color change.

[ Result ]

Result	No evidence of discoloration.
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## 8. Copper plate corrosion

Polish the surface of a copper plate of 0.3×30×30mm in size with metal abrasive, or polish and remove the oxide film with No. 1500 abrasive paper specified in JIS-R-6252 while bathed in organic solvent such as xylene, and after washing out the soil adhering to the surface with alcohol, etc., leave it in the air to dry completely.

Place the sample of approximately 0.1g on the copper plate, melt it by heating for about 5 sec. At 250°C and cool it at room temperature to obtain the test pieces.

Put three test pieces in a thermohygrostat of temperature 40°C×95%RH and humidity 95% for 96 hours and compare them with the reference test piece for the evidence of corrosion.

[ Result ]

n	1	No corrosion
	2	No corrosion
	3	No corrosion
Average		No corrosion

## 9. Copper mirror corrosion

This test method is designed to determine the removal effect the flux has (if any) on the bright copper mirror film which has been vacuum deposited on clear glass.

Apply by vacuum deposition, a film of copper metal on one surface of a cleaned glass sized 1.0×52×76mm specified in JIS-R-3703.

Apply a uniform thickness of approximately 50nm and assure that the finished mirror permits 10±5% transmission of normal incident light of nominal wave length of 500nm.

Place one drop of test flux on each copper mirror test panel.

Place test panels in a horizontal position in the dust free cabinet at  $23\pm 2^{\circ}\text{C}$  and  $50\pm 5\%$  relative humidity for 24 hours.

At the end of 24 hour period, remove the test panels and remove the test flux and control standard fluxes (isopropyl alcohol solution of 25wt% WW rosin) by isopropyl alcohol.

Carefully examine each test panel for possible copper removal or discoloration.

[ Result ]

	<b>JS-3000V-3 (2-propanol solution)</b>	WW rosin 25wt% I.P.A. solution
Result	Partial breakthrough	No breakthrough

## 10. Resistivity of water extract

Extract the flux in purified water and carry out the test on watersoluble conductive components in the flux measuring the conductivity of the extracted water at  $20^{\circ}\text{C}$ .

Take an amount of 0.1ml flux as the sample into a cleaned and dried 100ml beaker.

Put the sample in the beaker with 50ml of purified water, then cover the beaker with a watch glass, heat and boil it for about 5 minutes, and further continue heating for about 1 minute. Cool the beaker for about 10 seconds at room temperature, put it in a water bath of about  $20^{\circ}\text{C}$  to obtain the test solution, and immediately measure the resistivity of this water solution with a conductivity meter.

The cell of 0.1 cell constant shall be used.

The purified water to use shall have more than  $5 \times 10^4 \Omega \cdot \text{m}$  of specific resistance.

The test shall be made 3 times and take the mean value.

[ Result ]

N	1	$1.8 \times 10^2$
	2	$1.8 \times 10^2$
	3	$1.8 \times 10^2$
Average ( $\Omega \cdot \text{m}$ )		$1.8 \times 10^2$

\*Control standard (without flux) :  $5 \times 10^4 \Omega \cdot \text{m}$

## 11. Solder spreadability

Solder ring : Wind one turn in a ring form solder wire H60A-W1.6 specified in JIS-Z-3282 and S3X(Sn-3.0Ag-0.5Cu) around a bar with a diameter of 3.2mm to obtain the sample.

Test plate : Use as test plate a phosphor deoxidized copper plate specified in JIS-H-3100, 0.3×50×50mm in size polished by #1500 abrasive paper and washed by alcohol, subject it to oxidizing treatment in electric furnace maintained at about 150°C for 1 hour.

Test method : Place the test piece on the test plate and heat it at 250±5°C. Melt it for about 30 sec. After reaching the said temperature, spread the solder over the plate.

After cooling it at ordinary temperature, remove the residual flux with alcohol, and measure the height of solder and calculate the rate of spread from the following formula :

$$S = \frac{D - H}{D} \times 100$$

S : Rate of solder spreading ..... (%)  
 H : Height of spread solder ..... (mm)  
 D : Diameter when the solder used is assumed to be as sphere ..... (mm)

$$D = 1.2407V^{1/3}$$

V : Mass / specific gravity

[ Result ]

		Sn-37Pb	Sn-3.0Ag-0.5Cu
n	1	92.1	77.6
	2	92.5	77.5
	3	91.3	76.5
	4	91.7	76.7
	5	92.1	77.6
Average (%)		91.9	77.2

## 12. Insulation resistance

As a test piece, use the comb type electrode of the glass fiber-based copper-clad, epoxy resin GE-3 and GE-4, both specified in JIS-C-6480. After cleaning with alcohol and thoroughly drying the surface, uniformly apply a specific quantity (JIS type II = 0.05ml) of flux onto the electrode and solder at about 250°C for 4sec after dry it at about 100°C for 10min. solder a lead wire onto each terminal to obtain the test piece.

Prepare three pieces of the above test piece and measure the insulation resistance (initial value = DRY) under the above specified condition.

Put all the test pieces in a thermohygrostat and connect each lead wire with the terminals outside of the thermohygrostat.

Raise the temperature to a specific temperature first, then increase the humidity upto a specific

humidity.

After a specific time,

- (1) Measure the insulation resistance keeping the test pieces in the thermohygrostat
- (2) Take the test pieces out of the thermohygrostat, and measure the insulation resistance under the normal temperature and humidity.

Voltage to apply shall be DC100V.

Measurement shall be conducted at 4 points between each terminal pair per test piece and be expressed as a mean value.

\* Test conditions : 85°C×85%RH×168 hours

[ Result ]

		DRY = Soldering	In thermohygrostat	Out of thermohygrostat
N	1	$1.8 \times 10^{13}$	$7.0 \times 10^9$	$7.1 \times 10^{13}$
	2	$1.1 \times 10^{13}$	$5.3 \times 10^9$	$4.8 \times 10^{13}$
	3	$5.0 \times 10^{13}$	$1.1 \times 10^{10}$	$6.4 \times 10^{13}$
Average ( $\Omega$ )		$2.1 \times 10^{13}$	$7.4 \times 10^9$	$6.0 \times 10^{13}$

### 13. Voltage applied insulation resistance

As a test piece, use the comb type electrode of the glass fiber-based copper-clad, epoxy resin GE-3 and GE-4, both specified in JIS-C-6480. After cleaning with alcohol and thoroughly drying the surface, uniformly apply a specific quantity (JIS type II=0.05ml) of flux onto the electrode and solder at about 250°C for 4sec after dry it at about 100°C for 10min. solder a lead wire onto each terminal to obtain the test piece.

Prepare three pieces of the above test piece and measure the insulation resistance (initial value = DRY) under the above specified condition.

Put all the test pieces in a thermohygrostat and connect each lead wire with the terminals outside of the thermohygrostat.

Raise the temperature to a specific temperature first, then increase the humidity upto a specific humidity, and apply DC50V.

After a specific time,

- (1) Measure the insulation resistance keeping the test pieces in the thermohygrostat.
- (2) Take the test pieces out of the thermohygrostat, and measure the insulation resistance under the normal temperature and humidity.

Voltage to apply shall be DC100V for the measurement.

Measurement shall be conducted at 4 points between each terminal pair per test piece and be expressed as a mean value.

\* Test conditions : 85°C×85%RH×168 hours



[ Result ]

Sample		DRY = Soldering	In thermohygrostat	Out of thermohygrostat
n	1	$2.3 \times 10^{14}$	$5.3 \times 10^9$	$3.0 \times 10^{13}$
	2	$4.7 \times 10^{13}$	$1.0 \times 10^{10}$	$6.8 \times 10^{13}$
	3	$4.3 \times 10^{13}$	$1.2 \times 10^{10}$	$4.6 \times 10^{13}$
Average ( $\Omega$ )		$7.8 \times 10^{13}$	$8.7 \times 10^9$	$4.5 \times 10^{13}$

\* No evidence of electromigration nor corrosion.

#### 14. Wetting (Meniscograph)

Test plate : Use as test plate a phosphor deoxidized copper plate specified in JIS-H-3100, and nickel plate, 0.2×7×30mm in size polished by #1500 abrasive paper and washed by alcohol.

Condition : Solder temperature -  $250 \pm 2^\circ\text{C}$   
 Dipping depth - 2mm  
 Dipping speed - 25mm/min.  
 Solder quality - Pb Free Solder (Sn-3.0Ag-0.5Cu)

[ Result ]

Item		Samples			
		Polished copper plate		Polished nickel plate	
		Wetting speed (sec.)	Tensile strength $H_4$ mN/m	Wetting speed (sec.)	Tensile strength $H_4$ mN/m
n	1	0.73	393	0.72	302
	2	0.63	381	0.62	327
	3	0.50	392	0.72	321
	4	0.48	386	0.69	317
	5	0.54	387	0.68	307
Average		0.57	388	0.69	315

## 15. Dryness

After cleaning the surface of copper plate of 0.3×30×30mm with alcohol, etc. and drying, put 0.5g of the sample on the copper plate and melt it in about 5 sec. at a temperature of 250±5°C.

After leaving the test piece for 30 minutes at room temperature, sprinkle some white chalk powder on the surface of the remaining flux and check if the chalk powder can be removed by soft brushing.

[ Result ]

n	1	No attachment of chalk powder
	2	No attachment of chalk powder
	3	No attachment of chalk powder

## 16. Product handling

### 1) Storage

- Avoid water and direct sun, and store at room temperature.
- Shelf life is one year without opening, and three months once it is opened.
- Keep container tightly closed.
- Keep away from children.

### 2) Before use

- Use only with adequate ventilation.
- Wear adequate protective goggles and gloves.
- This No.0401 is for only 'spray' application.
- Make sure that a tank and pipeline of spray fluxing unit has been thoroughly cleaned.

### 3) Flux activation check

- Since No.0401 is strictly for 'spray' application and there is no chance of change of neither activation nor specific gravity, neither activation check nor adjustment shall be required.

### 4) Disposal

- The product is an water solution containing water, and acids, and may contain copper, tin and other heavy metals.  
Disposal procedure should follow local regulation and law.

## 17. Product use

### 1) Applicable PC boards

- Substrate : Paper phenol, glass epoxy, BT resin, ceramic, and etc.
- Surface treatment : HASL, OSP, rosin pre-flux and etc.
- Component : Leaded components, surface mount components

### 2) Fluxing

- This product is only for 'spray' application.
- Recommended amount to apply : 1.2 ~ 2.0 ml/100cm<sup>2</sup>

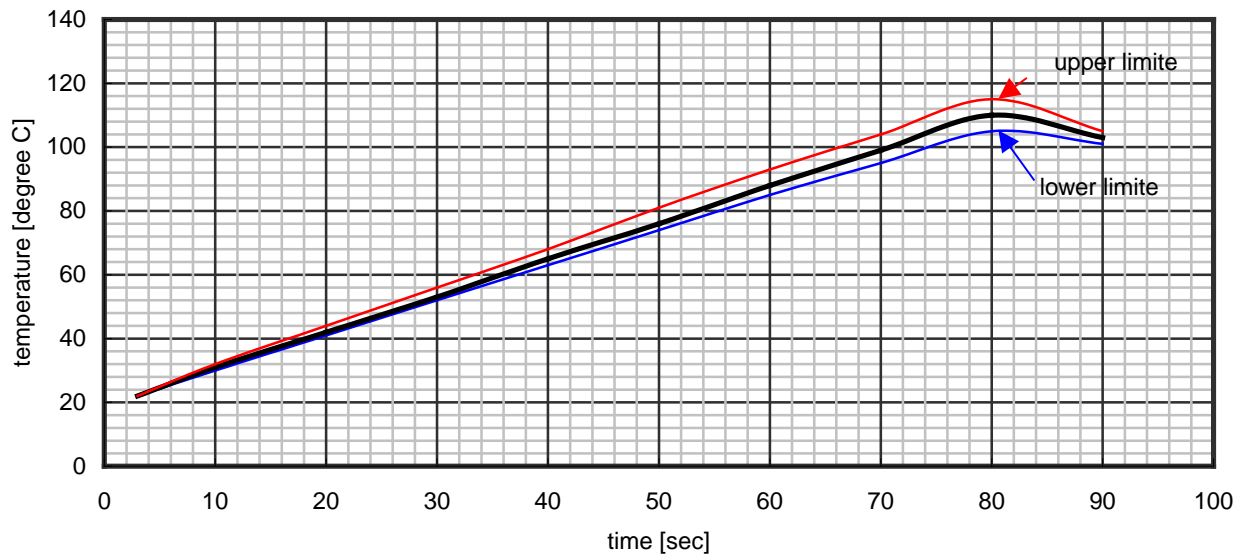
\*It is recommended to apply minimum needed amount by carefully observing actual soldering results.

### 3) Pre-heating

- Recommended pre-heat condition : 105 ~ 115°C at soldering side of PC board for 50 ~ 120 sec.  
 $\Delta T = 0.5 \sim 1.0^\circ\text{C}/\text{sec}$ .
- Flux activation temperature : approx. 85 ~ 260°C

\*Example :

Temperature profile for No.JS-3000V-3  
\* Soldering Side



### 4) Soldering

- Conveyor speed : 1.0 ~ 1.8m/min
- Conveyor angle : 3 ~ 6 degrees
- Solder temp. : 240 ~ 260
- Dip time. : 2 ~ 5 sec. (total time of first and 2nd wave)